

SUL ROSS STATE UNIVERSITY
Syllabus for General Chemistry I Lab: CHEM 1111: 1L1
CRN:30179
(Summer 2025)

LAB: General Chemistry 1111
Lab Section: 2L1
Room: WSB 307 or Online
Time: M, W, F
1:00-2:40 pm

TA: Dr. Hong Young Chang
Email: hxc19tv@sulross.edu
Office: WSB 219
Office Hours: After appointment via Zoom

Required Laboratory Manual

Laboratory Manual CHEM 1111 is uploaded to Blackboard. Students need to download and print out the corresponding lab manual for each lab experiment before class.

Expectations and Safety

- Read over the experiment before the lab
- Follow all safety procedures:
- Shorts, flip-flops, **chewing gum and open-toed shoes ARE NOT** allowed in the lab. If you come to class without appropriate clothing, you will be asked to leave. **NO EXCEPTIONS!**
- **Food and drink ARE NOT** allowed in the laboratory for your safety.
- **Safety glasses ARE REQUIRED** for General Chemistry laboratories, which can be purchased from the SRSU Bookstore. You will NOT be allowed to participate in General Chemistry laboratories without safety glasses! Note: prescription glasses count as safety glasses.

- Laboratory coat is also recommended for General Chemistry laboratories, which can be purchased from the SRSU Bookstore.
- If anyone is pregnant or gets pregnant during the semester, please inform your TA and Dr. Chang.
- **Calculator:** A scientific calculator is required for this course.
- **Cell phones ARE NOT** permitted for use in exams and should be turned off during laboratory time.

Attendance:

Coming to the lab is mandatory. Be on time and SIGN IN at the beginning of the lab period. Plan to spend the entire period in lab. The TA may deduct points for students who arrive late or leave early. **If you miss 3 labs or more, you will receive an automatic F.**

Assignments:

- **Pre-Lab:** Due to the beginning of the lab the experiment will be performed. **(It must be done in Blackboard before Lab experiment)**
- **Data Sheet:** Due to the lab period after the lab is done. **(Send it via email)**
- **Post-Lab Questions:** Due to the lab period after the lab is done.
(It has also to be done in Blackboard after Lab experiment)
- **Their due dates will be assigned in the course calendar.**

Lab Grading:

- Each experiment is worth 30 points
- These points will come from:
 - o The pre-lab definitions and Post-Lab Questions (20)
 - o Attendance (5)
 - o The experiment work & on time submission of data & results sheets/ written lab report (5)
 - o **Note:** Points will be deducted for not turning in lab reports and data & results sheets on time.
- Assignments must be completed and turned in on time (Their due date /time is announced in the course calendar.)

Scholastic Dishonesty:

Students who violate the University rules on scholastic dishonesty are subject to penalties, including the possibility of an **F** in the course and/or dismissal from the University. **All assignments and lab reports (including Pre-Lab) need to be individually completed and not copied from another student's work.**

Outline for Pre-lab and Post-lab Questions Solving:

- **Aim:** Here you will solve the questions related to the goal of the experiment
- **Reagents:** You will know a list of all the chemicals used in the experiment along with relevant data (grams, volume, molarity etc. that is indicated in your experiment). Thus, you will solve the questions related to chemicals and relevant data described above.

- **Apparatus:** You will know all the equipment that you will use. This equipment is also questioned in Pre-lab and post-lab questions.
- **Method:** This is where you will outline the steps in the experiment. The steps will be questioned as a multiple choice.

Outline for Written Lab Reports/Data & Results Sheet

(Use MS Word or related program)

- **Aim:** Here you will state the goal of the experiment (in your own words).
- **Reagents:** You will make a list of all the chemicals used in the experiment along with relevant data (grams, volume, molarity etc.). This is how much **YOU** used, not how much the manual asks for.
- **Apparatus:** You will list all the equipment that you used.
- **Method:** This is where you will outline the steps in the experiment. Be sure to note any difference between what you did and what the manual said to do.
- **Data & Results:** Note the observations that you made during the experiment. What are your findings? (Percent yield, melting point, etc.)
- **Discussion:** Discuss your results and answer the questions that were asked in the **Data and Results** section of the experiment. Talk about the significance of your results. Were your results expected or unexpected? Why or why not?
- **Conclusion(s):** Summarize the key points and findings of the experiment. Was the experiment successful or unsuccessful?

Only the Data & Results sheet must be submitted via email during this

summer semester. They must be submitted as a PDF file.

Learning Objectives and Outcomes of General Chemistry I Lecture and Lab:

At the end of this course, a student should have a good understanding of:

1. The basic concepts and terms used in chemistry: Metric System, Mole, Matter, Energy in chemical process, application of the conservation of mass law, Chemical Formula, Ions and Molecules, etc.
2. The electronic structures of atoms and the periodic table trends
3. The basic concepts of chemical bonding (Covalent Bonding and ionic Bonding)
4. Chemical reactions in aqueous solutions; Evidence of chemical reactions, how to write chemical equations, and how to balance chemical equations.
5. The ideal gas equation; relations between volume, mass, and pressure

Student Learning Outcomes (SLO) by SACSCOC:

A student graduating with a ***chemistry major*** is expected to demonstrate that (s)he can do the following:

1. Organic Chemistry—Students will be able to draw organic molecular structures and explain organic reactions, stereochemistry, structural analysis, and reactions in biological systems.
2. Inorganic Chemistry—The student will be able to demonstrate an understanding of coordination chemistry, valence theory, elementary actions, and advanced molecular theory.

3. Analytical Chemistry—The student will be able to demonstrate an understanding of the theory of analytical chemistry and conduct analytical analysis, including data analysis and calibration, equilibrium chemistry, gravimetric analysis, titrimetric analysis, spectroscopic analysis, and electrochemical analysis.

4. Physical Chemistry—The student will be able to demonstrate an understanding of the application and theory of physical chemistry, including topics such as atomic structure, electrochemistry, surface chemistry, solid-state chemistry, and thermodynamics.

5. Research—The student will collect and analyze published chemical literature and undertake a chemistry research project.

BSc in Chemistry Marketable Skills:

1. Students will become good at punctuality and time management.
2. Students will analyze &/or synthesize molecules and perform spectroscopic characterization and interpret their results scientifically.
3. Students will become proficient at writing scientific papers and identify appropriate references for their paper.
4. Students will become proficient at orally presenting scientific topics including the use of visual aids.

The SLOs as Core Curriculum Courses:

1. Critical Thinking Skills – Students will gain/improve their critical thinking ability by solving real-life chemistry problems through inquiry, analysis, and evaluation of available information. Students will be tested on their critical thinking ability in

exams and through lab experiments. In addition, students will have opportunities for simple self-assessments on critical thinking.

2. Communication Skills – Students will have the opportunity to improve communication skills through oral discussion and writing reports (i.e., observation, explanation, conclusion, etc.) on the experiments done in the lab sessions. In addition, students will have opportunities for simple self-assessments on communication skills.

3. Empirical and Quantitative Skills – Students will use the mathematical skills needed to manipulate and analyze numerical data obtained through experimentation in order to form conclusions.

4. Teamwork – Students will use team spirit and consider different points of view to work effectively while conducting experiments as a team working toward a shared purpose or goal.

SRSU Disability Services: ADA (Americans with Disabilities Act):

Sul Ross State University (SRSU) is committed to equal access in compliance with the Americans with Disabilities Act of 1973. It is SRSU policy to provide reasonable accommodations to students with documented disabilities. It is the student's responsibility to initiate a request each semester for each class. Students seeking accessibility/accommodations services must contact Mrs. Mary Schwartze Grisham, LPC, SRSU's Accessibility Services Director at 432-837-8203 or email mschwartz@sulross.edu. Our office is located on the first floor of Ferguson Hall, room 112, and our mailing address is P.O. Box C122, Sul Ross State University, Alpine, Texas, 79832.

Libraries:

The Bryan Wildenthal Memorial Library in Alpine offers FREE resources and services to the entire SRSU community. Access and borrow books, articles, and more by visiting the library's website, library.sulross.edu/. Off-campus access requires logging in with your LoboID and password. Librarians are a tremendous resource for your coursework and can be reached in person, by email (srsulibrary@sulross.edu), or by phone (432-837-8123).

No matter where you are based, public libraries and many academic and special libraries welcome the general public into their spaces for study. SRSU TexShare Cardholders can access additional services and resources at various libraries across Texas. Learn more about the TexShare program by visiting library.sulross.edu/find-and-borrow/texshare/ or ask a librarian by emailing srsulibrary@sulross.edu.

News for Fall 2024: Mike Fernandez, SRSU Librarian, is based in Eagle Pass (Building D-129) to offer specialized library services to students, faculty, and staff. Utilize free services such as Interlibrary Loan (ILL) and Scant to get materials delivered to you at home or via email.

Academic Integrity:

Students in this class are expected to demonstrate scholarly behavior and academic honesty in the use of intellectual property. Students should submit work that is their own and avoid the temptation to engage in behaviors that violate academic integrity, such as turning in work as original that was used in whole or part for another course and/or professor; turning in another person's work as one's own; copying from professional works or internet sites without citation; collaborating on a course assignment, examination, or quiz when collaboration is forbidden.

Students should also avoid using open AI sources ***unless permission is expressly given*** for an assignment or course. Violations of academic integrity can result in failing assignments, failing a class, and/or more serious university consequences. These behaviors also erode the value of college degrees and higher education overall.

Counselling:

Sul Ross has partnered with TimelyCare where all SR students will have access to nine free counseling sessions. You can learn more about this 24/7/365 support by visiting [Timelycare/SRSU](#). The SR Counseling and Accessibility Services office will continue to offer in-person counseling in Ferguson Hall room 112 (Alpine campus), and telehealth Zoom sessions for remote students and RGC students.

Classroom Climate of Respect:

Importantly, this class will foster free expression, critical investigation, and the open discussion of ideas. This means that all of us must help create and sustain an atmosphere of tolerance, civility, and respect for the viewpoints of others. Similarly, we must all learn how to probe, oppose and disagree without resorting to tactics of intimidation, harassment, or personal attack. No one is entitled to harass, belittle, or discriminate against another based on race, religion, ethnicity, age, gender, national origin, or sexual preference. Still, we will not be silenced by the difficulty of fruitfully discussing politically sensitive issues.

Distance Education:

Students should correspond using Sul Ross email accounts and submit online assignments through Blackboard, which requires a secure login. Students enrolled

in distance education courses at Sul Ross are expected to adhere to all policies pertaining to academic honesty and appropriate student conduct, as described in the student handbook. Students on web-based courses must maintain appropriate equipment and software, according to the needs and requirements of the course, as outlined on the SRSU website. Directions for filing student complaints are in the student handbook.

General Chemistry 1111 Lab Schedule and Brief Description (Course Calendar)

This is the course calendar. You need to focus on the date, experimental chapter number, topics, class type, and final examination day. This course calendar could be changed. Before one week, your professor will let you know the changes.

Date	Experiments	Pre-Lab	Data Results	Post-Lab
28-May	No Laboratory Experiments	x	x	x
30-May	Safety Video Watching, Class Introduction, Safety Practices and Procedures in the Laboratory (Exp.1)	x	x	0
2-Jun	Chemistry Laboratory Equipment (Exp.1_2)	x	x	0
4-Jun	Chemical Calculations and Symbols (Exp. 2)	0	0	0
6-Jun	Naming of Chemical Compounds (Exp. 3)	0	0	0
9-Jun	Balancing Chemical Equations (Exp. 4)	0	0	0
11-Jun	Exercise of Mole Calculation (Exp. 5)	0	0	0
13-Jun	Exercise of Mole Calculation (Exp. 6)	0	0	0
16-Jun	Metathesis Reactions (Exp. 7)	0	0	0
25-Jun	Online Final Exam (in Blackboard)			