# **SYLLABUS BIOCHEMISTRY**

(2025\_FALL\_12525\_CHEM\_4301 001\_Biochemistry I)

**Start Date:** 08/22/2025 **End Date:** 12/7/2025

Type: Classroom

**Building:** Warnock Science **Room**:00307

Class hours: Tuesday/Thursday 09:30AM-10:45PM



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**Contacting Me:** The best way to reach me is via email. Do not hesitate to contact me for any reason. I will respond to email inquiries as soon as possible.

### **COURSE MATERIALS**

The following course materials are required

- 1. David L. Nelson; Michael M. Cox. Eighth Edition, 2021. Lehninger Principles of Biochemistry. Macmillan Learning. [VitalSource Bookshelf version] ISBN: 9781319350147
- 2. Recommended-American Psychological Association (October 2019). Publication Manual of the American Psychological Association, Seventh Edition, <a href="https://apastyle.apa.org/products/publication-manual-7th-edition-spiral">https://apastyle.apa.org/products/publication-manual-7th-edition-spiral</a>

Additional resources, including supplementary readings, videos, and other materials, will be made available on Blackboard throughout the semester.

## **SRSU Library Services and Support**

The Bryan Wildenthal Memorial Library at Sul Ross State University is your go-to hub for academic resources, offering a wealth of materials and services to the entire SRSU community at no cost. You'll find a vast collection of books, articles, and online databases readily accessible at <a href="https://www.library.sulross.edu">www.library.sulross.edu</a>. If you're accessing resources off-campus, simply use your Lobo ID and password to log in. For checking out physical materials, remember to bring a photo ID.

Need assistance? Our librarians are here to help! You can reach them via email at srsulibrary@sulross.edu, by phone at 432-837-8123, or by visiting them in person.

## **Course Description**

Biochemistry I is an upper-level undergraduate course that introduces the chemical principles underlying biological molecules and processes. The course covers the structure, properties, and functions of the major classes of biomolecules, including amino acids, proteins, carbohydrates, nucleotides, nucleic acids, and lipids. Topics also include the role of water in biological systems, protein structure and function, enzyme kinetics and mechanisms, glycobiology, membrane structure and transport, and the principles of bioenergetics.

Emphasis is placed on developing critical thinking and problem-solving skills by applying chemical concepts to biological systems. This course provides the foundation for advanced studies in metabolism, molecular biology, biotechnology, and related disciplines

### **Student Learning Objectives**

Upon successful completion of this course, students will be able to:

- Foundations of Biochemistry: Explain the scope of biochemistry, cellular organization, and the chemical basis of life.
- Water & Noncovalent Interactions: Describe the properties of water, pH, buffering systems, and the role of weak interactions in biomolecular stability.
- Amino Acids & Proteins: Identify amino acid structures and properties; explain peptide bond formation, protein purification, and characterization techniques.
- **Protein Function**: Analyze the relationship between protein structure and function, including ligand binding, enzyme activity, and oxygen transport.
- Enzymes: Apply principles of enzyme kinetics, inhibition, and catalytic mechanisms.
- Carbohydrates: Classify carbohydrates and glycoconjugates and explain their structural and biological roles.
- Nucleotides & Nucleic Acids: Describe the structure, chemical properties, and biological functions of nucleotides, DNA, and RNA.
- Lipids & Membranes: Explain lipid structure, function, and classification; evaluate the organization and properties of biological membranes.
- Membrane Transport: Differentiate between passive and active transport and analyze the role of membrane proteins and ion channels.

## **Student Learning Outcomes (SLO)**

A student graduating with a chemistry major is expected to demonstrate that (s)he is able to do the following:

- 1. Organic Chemistry—Students will be able to draw organic molecular structures and explain organic reactions, stereochemistry, structural analysis, and reactions in biological systems.
- 2. Inorganic Chemistry—The student will be able to demonstrate an understanding of coordination chemistry, valence theory, elementary actions, and advanced molecular theory.
- 3. Analytical Chemistry—The student will be able to demonstrate an understanding of the theory of analytical chemistry and conduct analytical analysis, including data analysis and calibration, equilibrium chemistry, gravimetric analysis, titrimetric analysis, and spectroscopic analysis, and electrochemical analysis.
- 4. Physical Chemistry—The student will be able to demonstrate an understanding of the application and theory of physical chemistry, including topics such as atomic structure, electrochemistry, surface chemistry, solid-state chemistry, and thermodynamics.
- 5. Research—The student will collect and analyze published chemical literature and undertake a chemistry research project.

### **BSc in Chemistry Marketable Skills**

- 1. Students will become good at punctuality and time management.
- 2. Students will analyze &/or synthesize molecules and perform spectroscopic characterization and interpret their results scientifically.
- 3. Students will become proficient at writing scientific papers and identifying appropriate references for the papers.
- 4. Students will become proficient at orally presenting scientific topics including the use of visual aids.

### **Technology Requirements**

This course requires frequent use of **Blackboard**. Students must check it regularly for announcements, assignments, grades, and course materials. Key Blackboard tools include email, course documents, discussion boards, grade center, SafeAssign, and external links.

# **Basic Computer Skills Needed:**

- Sending/receiving email with attachments
- Browsing the internet and downloading files
- Using word processing software (e.g., MS Word) **Required Technology:**
- A reliable computer with stable internet access
- A current web browser capable of streaming content and downloading materials
- Access to Blackboard using your Lobo ID and password
- For assistance, tutorials are available on the Blackboard support page.

## **Participation Policy**

This is a fully online course with no class meetings scheduled. However, active participation is expected. To start, please complete the non-graded "Introduction" post to introduce yourself to your classmates and your professor.

Students are expected to:

- Log into Blackboard regularly to check announcements, assignments, and deadlines
- Complete all assignments and exams online by their posted due dates
- Contact the instructor via email with any questions about course materials, assignments, or exams.
- Consistent engagement is key to success in this course.

#### **Attendance Policy**

Although this is a fully online course, student engagement and participation are still mandatory. The University's attendance policy applies to all distance learning courses. A student may be dropped with an "F" grade for excessive non-participation—defined as more than 3 weeks of inactivity in a long semester, 1 week in a summer session, or 3 days in a midwinter session.

#### **Inactivity includes:**

- Not logging into Blackboard
- Not submitting assignments
- Not participating in required activities
- Not communicating with the instructor
- Not following the participation guidelines in the syllabus

Students must respect academic integrity and copyright laws. All submitted work must be their own unless group work is specifically assigned. Additionally, students accessing the course from a remote site are expected to treat the host campus, its facilities, and staff with professionalism and care.

Students are required to check their **SRSU** email accounts regularly, as all official communications—including those sent via Blackboard—will be directed there.

## **Late Assignment Submission Policy**

All coursework and assignments must be submitted by the stated deadlines. Late work is generally not accepted, as students are informed of all due dates at the beginning of the term and are expected to plan accordingly.

Exceptions will only be made in cases of **documented and university-recognized excuses**. If a student misses an assignment or exam, they must promptly provide a valid explanation and supporting documentation.

Students are strongly encouraged to review the syllabus regularly and stay informed of all deadlines throughout the semester.

#### **Academic Integrity**

Students are expected to uphold the highest standards of academic honesty as outlined in the **Sul Ross Student Code of Conduct**. Any form of academic dishonesty—including **cheating**, **plagiarism**, **collusion**, **fabrication**, **or falsification of records**—will not be tolerated and will be addressed in accordance with university policies.

#### **Definitions:**

- **Cheating:** Using unauthorized materials or assistance during exams, altering academic records, or obtaining answers through dishonest means.
- Plagiarism: Submitting another's work, ideas, or words as your own without proper citation.
- Collusion: Assisting or being complicit in another student's academic dishonesty.

Using tools like **ChatGPT** or other **AI** generators is prohibited unless explicitly permitted for a specific assignment. Violations may result in penalties ranging from a failing grade on an assignment to failure of the course and may lead to further disciplinary action.

Maintaining academic integrity protects both the value of your education and the credibility of the university.

# **Student Support Services**

Sul Ross State University offers various programs to help students succeed, including advising, counseling, mentoring, tutoring, supplemental instruction, and writing assistance. For a complete list of services, visit Student Support Services at <a href="https://www.sulross.edu/section/311/student-support-services">https://www.sulross.edu/section/311/student-support-services</a>. For more information, contact SSS at (432) 837-9118 or visit Ferguson Hall, Room 105.

**Counseling Services**: Sul Ross students have access to nine free counseling sessions through TimelyCare. For 24/7 support, visit TimelyCare/SRSU. In-person counseling is also available in Ferguson Hall, Room 112 (Alpine campus), and via telehealth for remote and RGC students.

### Students with Special Needs - Americans with Disabilities Act as Amended (ADAAA)

Sul Ross State University (SRSU) is committed to equal access in compliance with Americans with Disabilities Act of 1973. It is SRSU policy to provide reasonable accommodations to students with documented disabilities. It is the student's responsibility to initiate a request each semester for each class. Alpine students seeking accessibility/accommodations services must contact Mary Schwartze Grisham, M.Ed., LPC, SRSU's Accessibility Services Coordinator at

432-837-8203, or email <u>mschwartze@sulross.edu</u> The office is located on the first floor of Ferguson Hall # 112, and the mailing address is P.O. Box C-122, SRSU, Alpine. Texas, 79832.

#### **Course Evaluation**

- Attendance Requirement: Students who are absent for more than nine (9) lectures will automatically fail the course.
- Classroom Policy: Cell phones must be turned off during class and are strictly prohibited during examinations.
- Homework (30%): Homework will be assigned throughout the semester and will account for 30% of the final grade. All assignments must be completed in pen and submitted by the specified due date.
- Examinations: The course includes one midterm examination and a comprehensive final examination. No make-up exams will be offered. The final exam is mandatory.

• Important Dates: Students are responsible for consulting the course calendar and the university's academic calendar on Blackboard for all due dates and deadlines.

# Grading

# **Grading Components**

Assessment	%
Homework	30
Midterm Exam	30
Final Exam	40
Total Possible Points	100

## Course Grade Scale

Letter Grade	Points Range
A	100 –90
В	89 – 80
С	79 – 60
D	59 –50
F	49 & below

### **Course Schedule**

This course schedule is subject to change as needed to meet the needs of the course. Students will be notified of adjustments when they are made. Also, any additional reading materials, resources and other information will be posted on Blackboard.

Students will be notified of how to access this information by the instructor. \*\*\*All times are in central time\*\*\*

Week	Dates	Topics	Assignments
1	August 25-31	✓ Course Introduction & Requirements	Syllabus & Semester Project Guidelines
		<ul> <li>✓ Cellular Foundations</li> <li>✓ Chemical Foundations</li> <li>✓ Physical Foundations</li> <li>✓ Genetic Foundations</li> <li>✓ Evolutionary Foundations</li> </ul>	Chapter 1: Foundations of Biochemistry
2	September 1-7 Sep 1 <sup>st</sup> , Labor Day No class	<ul> <li>✓ Weak Interactions in Aqueous System</li> <li>✓ Ionization of Water, Weak Acids, and Weak Bases</li> <li>✓ Buffering against pH Changes in Biological System</li> </ul>	Chapter 2: Water, The Solvent of Life
3	September 8-14	<ul><li>✓ Amino acids</li><li>✓ Peptides and Proteins</li><li>✓ Working with Proteins</li></ul>	Chapter 3:  Amino acids, Peptides, and Proteins

4	September 15-21	<ul><li>✓ The Structure of Proteins: Primary Structure</li><li>✓ Overview of Protein Structure</li></ul>	Chapter 3: Chapter 4:
5	September 22-28	<ul> <li>✓ Protein Secondary Structure</li> <li>✓ Protein Tertiary and Quaternary Structures</li> <li>✓ Protein Denaturation and Folding</li> <li>✓ Determination of protein and Biomolecular Structures</li> </ul>	Chapter 4:  The Three-Dimensional Structure of Proteins
6	Sep 29-Oct 5	<ul> <li>✓ Reversible Binding of a Protein to a Ligand:         Oxygen-Binding Proteins</li> <li>✓ Complementary Interactions between         Proteins and Ligands: The Immune System         and Immunoglobulins</li> <li>✓ Protein Interactions Modulated by Chemical         Energy</li> </ul>	Chapter 5: Protein Function
7	October 6-12	<ul> <li>✓ An Introduction to Enzymes</li> <li>✓ How Enzymes Work</li> <li>✓ Enzyme Kinetics are Approach to Understanding Mechanism</li> </ul>	Chapter 6: Enzymes
8	October 13-19 October 17, Final Day for Midterm Exam	✓ Midterm Exam	
9	October 20-26	<ul><li>✓ Examples of Enzymatic Reactions</li><li>✓ Regulatory Enzymes</li></ul>	Chapter 6:
10	October 27- Nov 2	<ul> <li>✓ Monosaccharides and Disaccharides</li> <li>✓ Polysaccharides</li> <li>✓ Glycoconjugates: Proteoglycans,         Glycoproteins, and Glycolipids</li> <li>✓ Carbohydrates as Informational Molecules:         The Sugar Code</li> <li>✓ Working with Carbohydrates</li> </ul>	Chapter 7: Carbohydrates and Glycobiology
11	November 3-9	<ul> <li>✓ -Some Basic Definitions and Conventions</li> <li>✓ -Nucleic Acid Structure</li> <li>✓ -Nucleic Acid Chemistry</li> <li>✓ -Other Functions of Nucleotides</li> </ul>	Chapter 8: Nucleotides and Nucleic Acids

12	November 10-16	<ul> <li>✓ -Storage Lipids</li> <li>✓ -Structural Lipids in Membranes</li> <li>✓ -Lipids as Signals, Cofactors, and Pigments</li> <li>✓ -Working with Lipids</li> </ul>	Chapter 10: Lipids
13	November 17-23	<ul> <li>✓ The Composition and Architecture of Membranes</li> <li>✓ Membrane Dynamics</li> <li>✓ Solute Transport Across Membranes</li> </ul>	Chapter 11:  Biological Membranes and Transport
14	November 24-Nov 30 Thanksgiving Day Holiday (11 / 26-28)	<ul> <li>✓ Biochemical Signaling</li> <li>✓ November 25<sup>th</sup>, the last day before Thanksgiving.</li> </ul>	Chapter 12: Biochemical Signaling
15	December 1-7 December 4	<ul><li>✓ Biochemical Signaling</li><li>✓ Last Class Day before Finals</li></ul>	Chapter 12: Biochemical Signaling
15	December 5-7	✓ Final Exams	

**END OF COURSE EVALUATIONS:** Student evaluations of faculty are administered online at the end of each term/session for all courses with five or more students. Students will receive an email containing a link to a survey for each course in which they are enrolled. All responses are anonymous.